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Upgrade of a Tail-End Acetylene Converter
BASF FINA Petrochemicals Limited Partnership Naphtha Cracker

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ABSTRACT

This paper will provide an overview on improving the performance of a tail-end acetylene converter through the implementation of a succession of changes. The converter is in a naphtha cracker located at Port Arthur, Texas. This cracker is a joint venture between BASF Corporation (“BASF”) and TOTAL PETROCHEMICALS USA, INC. (“TOTAL”). The operating entity is BASF FINA Petrochemicals Limited Partnership (“BFLP”). ABB Lummus Global, Inc. (“ABB”) designed and constructed the naphtha cracker complex.

During the initial startup phase and the early years of the cracker operation, the acetylene converter was a major factor in poor plant reliability and associated flaring. The flaring and production interruptions were quite costly.

The focus and thrust in dealing with the poor-performing acetylene converter could be characterized as “real-time, operations-oriented”. That is, the over-riding priority was to fix the problem. Since there was no obvious root cause, a multi-pronged approach was taken. Hence, this paper is directed towards the following:

- Describing the converter system and catalyst as installed
- Reviewing converter performance during the first two years of plant operation,
- Discussing the “fixes”: Some of the “fixes” didn’t address the problem; some did.
- Providing an overview description of the current acetylene converter system, after implementation of all the changes.
- Illustrating the good performance being achieved with the modified system.
INTRODUCTION

General Plant Information

The plant site is known as the “NAFTA Region Olefins Complex” (NROC). Figures 1 and 2 provide aerial views of the site. This site is in Port Arthur, Texas, at the mouth of the Neches River where it empties into Sabine Lake. Two distinct legal entities are located at the site: BASF FINA Petrochemicals Limited Partnership (BFLP) and Sabina Petrochemicals LLC (Sabina).

The BFLP entity encompasses the naphtha cracker, an on-purpose propylene unit, and an aromatics recovery unit. This is a joint venture between BASF and TOTAL.

Sabina is a C4-upgrade complex. This is a joint venture between Shell Chemicals, BASF, and TOTAL. Both Sabina and BFLP are operated by BASF. The site has only one Control Room. The plant site is situated adjacent to the TOTAL Port Arthur refinery.

Design capacities for the cracker primary products are: 2.1 Billion lbs./yr (950 KTA) of ethylene and 1.3 Billion lbs./yr (590 KTA) of propylene. The on-purpose propylene capacity is 690 Million lbs/yr. On-purpose propylene is produced via the metathesis route using ABB’s Olefins Conversion Technology (OCT). The aromatics recovery unit is a combination of BASF’s SELOP Pygas Hydrogenation Technology and Krupp Uhde’s MORPHYLANE Extractive Distillation Technology.

Butadiene is recovered via BASF’s NMP Technology. The capacity of the butadiene unit is 900 Million lbs/yr, making it the world’s largest single-train butadiene plant. Isobutylene and butene-1 are upgraded to a high-octane gasoline component via UOP’s InAlk Technology.

TOTAL’s refinery manages the logistics required for furnace feeds and cracker liquid products; e.g., benzene, toluene, and C8+ material.

The initial startup of the cracker got underway during November/December 2001.

Simplified Plant Schematic and Design Information on the Acetylene Converter.

Figure 3 is a simplified Block Diagram for the Cracker, illustrating location of the tail-end acetylene converter on the O/H of the Deethanizer.

Figure 4 illustrates the ABB acetylene converter system, at start-up. The acetylene converter was designed with two-beds in series. There is inter-cooling between the beds. Each two-bed reactor is contained in one vessel. The reactor is spared; that is, a two-bed reactor is online while the spare reactor is being regenerated or in standby.
Some design parameters are as follows:

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Specification</th>
</tr>
</thead>
<tbody>
<tr>
<td>Inlet Acetylene</td>
<td>1.1-1.3 mole %</td>
</tr>
<tr>
<td>Ethylene in Converter Feed</td>
<td>76%</td>
</tr>
<tr>
<td>Acetylene Specification for ethylene product</td>
<td>1.4 wppm (Max)</td>
</tr>
<tr>
<td>Maximum Acetylene Allowed in Converter Effluent</td>
<td>0.8 ppm</td>
</tr>
<tr>
<td>Bed Space velocity</td>
<td>5000 V/V/Hr</td>
</tr>
<tr>
<td>Bed L/D</td>
<td>0.7</td>
</tr>
<tr>
<td>Catalyst</td>
<td>Sud Chemie’s G-58D</td>
</tr>
<tr>
<td>CO in Feed</td>
<td>None</td>
</tr>
<tr>
<td>H2</td>
<td>Cracker Methanated H2: About 95.7 mole %</td>
</tr>
<tr>
<td></td>
<td>Backup pipeline H2 is available at 99.9 % purity</td>
</tr>
<tr>
<td>Regeneration</td>
<td>In situ steam/air</td>
</tr>
</tbody>
</table>

The acetylene converter was designed with a catalyst selection basis that did not require CO moderation. The catalyst selected was Sud Chemie’s G-58D. This selection was based on information available about industrial experience on operating G-58D in tail-end acetylene converters without CO moderation.

The ethylene product has a CO specification of 100 mppb, maximum, and the C2 fractionator has a pasteurization section; so, it is possible to safely inject CO into the feed of the converter at levels below about 2 mppm. However, during the design stage, it was decided to forego the installation CO injection facilities.

G-58D is a silver-promoted Pd-on-alumina catalyst. It is one of several versions of Ag/Pd acetylene converter catalysts offered by Sud Chemie. One of its primary attributes is that it does not require CO moderation for selectivity and stability maintenance. Some of the characteristics of G-58D are given below:

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Specification</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sud Chemie’s Designation</td>
<td>G-58D</td>
</tr>
<tr>
<td>Shape/Size</td>
<td>Tablets: 4 x 4 mm</td>
</tr>
<tr>
<td>Pd Content, wt %</td>
<td>0.018</td>
</tr>
<tr>
<td>Ag Content, wt%</td>
<td>0.120</td>
</tr>
<tr>
<td>Support Material</td>
<td>Alpha Alumina,</td>
</tr>
<tr>
<td>Surface Area, m²/gm</td>
<td>3-7</td>
</tr>
<tr>
<td>Bulk Density, Lbs/Ft³</td>
<td>72</td>
</tr>
</tbody>
</table>
The acetylene converter system was installed with 2 online analyzers. These analyzers monitored acetylene, H2 and ethane in the converter feed, at the mid-bed location, and in the converter effluent. The feed and mid-bed streams were monitored with a single analyzer, set up with automatic “stream switching” such that the amount of acetylene, H2, and ethane in each stream was updated every 4 minutes. The effluent analyzer provided updated component readouts every 5 minutes.

The design control strategy was to maintain the moles of H2 being fed to each bed by ratio-controlling off moles of acetylene in the feed to each bed. The operators reset these ratios as needed to maintain the acetylene in the converter effluent below 0.8 ppm.

OVERVIEW ON ACETYLENE CONVERTER SITUATION DURING 2002/2003---PERFORMANCE & TROUBLESHOOTING ACTIVITIES

In a nutshell, the performance of the acetylene converter during 2002 ranged from “disappointing to poor”. There were many acetylene off-spec incidents caused by the poor performance of the converter. Each off-spec incident resulted in flaring.

A summary of the poor performance of the converter during 2002 is as follows. This information was extracted from a presentation given to the TCEQ in September 2002.

- Number of Switchovers between R 4101 A and B: 7
- Average run length: 41 days
- Overall average selectivity: about “negative 30%”
- No of acetylene off-spec incidents: 31
- Estimated economic impact: Approx. $13M
  - Flaring losses
  - Losses from “expected selectivity”
  - Impact on recycle cracking furnace
- Highly unstable operation:
  - Flaring at each reactor switch
  - Operators “nightmare”
A team effort was directed towards this problem. This included support from BASF, TOTAL Petrochemicals, ABB, and Sud Chemie. There were a number of factors considered as being the “root cause” or major contributors to the problem. These include:

- Feed poisons
- Insufficient catalyst bed height; i.e., L/D should be 1.0 (min.) vs actual of 0.7
- Inadequate space velocity (at times, perhaps too low or too high)
- Catalyst not the “right choice” for this cracker.
- Inadequate control scheme and/or inadequate online analyzers
- Incorrect operating conditions – partially due to inexperience of operators
- Inadequate “reactor swapping procedures”
- Mal-distribution of flow through the catalyst beds
- Excessive green oil formation
- Problems with regeneration

In the “final analysis, there was no “single smoking gun” identified and fixed. However, the performance of the acetylene converter was dramatically improved by doing tasks to eliminate some of the proposed contributors to the problem and by implementing changes in facilities, catalyst, and operations. The following steps are the ones that resulted in the improved performance of the converter:

- Replacing the G-58D catalyst with Sud Chemie’s G-58C
- Installing facilities to inject CO in the converter feed (This is for “moderating G-58C)
- Installing a common “trim” reactor. This reactor is loaded with G-58C
- Analyzer upgrade: In general, additional analyzers were installed and these analyzers had shorter cycle times.
- Some procedures were changed and more training was given to the operators.
- The operators became “more experienced”

Improved converter performance began to be realized in late 2003 and early 2004. The remainder of this paper will cover the following:

- Historical data from the “poor performance period” (all of 2002 and a major portion of 2003)
- Comments on some of the diagnostic/troubleshooting work done in 2002/2003
- Results from some “performance improvement trials and activities”
- Description of current converter system
Overview on Converter Performance -- Early Cycles in 1st half of 2002

The performance of the converter during the first four cycles is outlined as follows:

<table>
<thead>
<tr>
<th>Cycle Number</th>
<th>Reactor A or B - Catalyst (G-58D)</th>
<th>SOR</th>
<th>EOR</th>
<th>Run length (days)</th>
<th>Overall average ethylene gain (Lbs/Hr)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1st A – G-58D</td>
<td>11/19/01</td>
<td>1/22/02</td>
<td>38</td>
<td></td>
<td>-484</td>
</tr>
<tr>
<td>1st B – G-58D</td>
<td>1/23/02</td>
<td>2/19/03</td>
<td>22</td>
<td></td>
<td>-215</td>
</tr>
<tr>
<td>2nd A – G-58D</td>
<td>2/20/02</td>
<td>3/19/03</td>
<td>26</td>
<td></td>
<td>-322</td>
</tr>
<tr>
<td>2nd B – G-58D</td>
<td>3/20/02</td>
<td>4/29/02</td>
<td>37</td>
<td></td>
<td>-1240</td>
</tr>
</tbody>
</table>

The operation of the converter during this period can be characterized by the following comments:

- Difficult to maintain consistent/stable operation during these cycles.
- Many episodes of flaring due to excessive acetylene in the converter effluent.
- Reactors could not be switched without an off-spec flaring incident.

Figures 8 through 11 show the ethylene gain trend during each of these cycles. These figures provide an illustration of the performance of the reactors during this period. There was no discernible difference between the performance of the “A” reactor and the “B” reactor.

Analyzer Upgrade Project Initiated

By the beginning May, 2002, the Converter Improvement Team had not been able to define a “Root Cause” for the poor performance of the converter. However, a portion of the problem was attributed to the long cycle times for the two online analyzers. For illustration, the analyzer cycle times were so long that the operators could not use molar H2/acetylene control. They could only use mass ratios of H2 flows/converter-feed to control H2 injected into each bed. This meant that the H2 injection rates were based on flow meters, not on analyzer readings. Also, acetylene breakthrough from the 2nd bed was not detected before the C2 fractionator was contaminated with acetylene. This resulted in much longer flaring durations before “on-spec” ethylene was re-established.

New analyzers were purchased for installation during the 2nd half of 2002. The analyzer upgrade project provided for the following analyzers:

- Two new acetylene analyzers on the outlet of the converter. The cycle time for each analyzer is 45 seconds.
- Two new analyzers at the mid bed locations: One analyzer on the outlet of the 1st Bed and one analyzer on the feed to the 2nd Bed. Each analyzer gave measurements for acetylene, H2, and ethane.
- An existing analyzer was reconfigured to monitor only the feed to the reactor. This analyzer gave faster updates for acetylene, H2, and ethane.
**Implemented Improved Dryout Procedure following Steam/Air Regeneration:**

Free water was discovered in the piping of a freshly regenerated reactor; so the dryout procedure was revised. This included using moisture analyzers at several locations to assure that Dew Points < -60 F are achieved during the dryout. Following this, there were no more incidents of excessive moisture following completion of the dryout procedure. However, this did not contribute to a noticeable “short-term” improvement in the performance of the converter.

**G-58D Catalyst in 2nd Bed of “A Reactor” Replaced with G-58C/C-31-3 -- May, 2002**

The catalyst in the 2nd bed of the “A” reactor was replaced with G-58C and C-31-3. The new bed configuration was as follows:

- Top Layer (75% of bed): G-58C
- Bottom Layer (25% of bed): C-31-3

This was done before returning the “A” reactor for the 3rd Cycle. A schematic of the reactor system with G-58C/C-31-1 loaded in the 2nd bed of the “A” reactor is shown in Figure 5.

This change did not bring about a significant improvement in the converter performance. This can be seen from the Ethylene Gain Profile shown in Figure 12. The cycle only lasted about 7 weeks and the ethylene gain decreased below “zero” in less than a week.

With the benefit of “hindsight”, it’s easy to say that this was not a “good step”. Nonetheless, here are some of the reasons, at the time, which led to the decision to install G-58C/C-31-3 in a 2nd bed.

- There was an “urgency to do something” that might improve converter performance
- The 2nd bed was where the instabilities were most visible with a direct impact on product quality.
  - This bed was most difficult to control. It was very unstable after only a few days online.
  - There were problems with excessive H2 breakthrough, coincident with acetylene breakthroughs
  - Some “thinking” that G-58D was not the “right catalyst” for this plant operation.
For reference, here are some properties for G-58C and C-31-3:

<table>
<thead>
<tr>
<th></th>
<th>G-58C</th>
<th>C-31-3</th>
</tr>
</thead>
<tbody>
<tr>
<td>Pd Content, wt%</td>
<td>0.03</td>
<td>0.14</td>
</tr>
<tr>
<td>Ag Content, wt%</td>
<td>0.18</td>
<td>None</td>
</tr>
<tr>
<td>Support</td>
<td>Alumina</td>
<td>Alumina</td>
</tr>
<tr>
<td>Particle Size/Shape</td>
<td>2-4 mm spheres</td>
<td>3-7 mm spheres</td>
</tr>
<tr>
<td>Bulk Density, Lbs/Ft³</td>
<td>44</td>
<td>44</td>
</tr>
</tbody>
</table>

Reasons for installing G-58C include:

- Good experience with this catalyst within TOTAL.
- Input from “outside companies” on good experience with G-58C.

Concerns over installing this bed configuration:

- No CO injection for moderation of the selectivity/stability of G-58C and C-31-3
- Insufficient length of G-58C “selective layer”: L/D of about 0.55
- C-31-3 is very non-selective in this environment. Dominant reaction will be complete consumption of H₂ via hydrogenation of ethylene. Hence, H₂ will not be available for “removal of trace acetylene”.

Other work done in the 2nd Half of 2002 eliminated some “proposed causes” for the poor performance of the converter.

This work is outlined as follows:

- Initial work by TOTAL on excessive green oil formation by G-58D was inconclusive. Corrective changes were made to the converter system before this work was completed.
- Samples of used G-58D removed from the 2nd bed of the “A” reactor were analyzed. The results from these analyses are given below. This work did not reveal noteworthy catalyst poisoning and/or catalyst damage. The only item “of note” is the higher level of iron found at the bottom of the 2nd bed.
Analytical Results from Used G-58D (April 2002)

<table>
<thead>
<tr>
<th>Species</th>
<th>Units</th>
<th>1st Bed</th>
<th>1st Bed</th>
<th>2nd Bed</th>
<th>2nd Bed</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>Top Layer</td>
<td>Btm Layer</td>
<td>Top Layer</td>
<td>Btm Layer</td>
</tr>
<tr>
<td>Pb</td>
<td>wppm</td>
<td>&lt;3</td>
<td>&lt;3</td>
<td>&lt;3</td>
<td>&lt;3</td>
</tr>
<tr>
<td>Ca</td>
<td>wppm</td>
<td>&lt;39</td>
<td>45</td>
<td>40</td>
<td>40</td>
</tr>
<tr>
<td>Cl</td>
<td>wppm</td>
<td>&lt;20</td>
<td>&lt;20</td>
<td>&lt;20</td>
<td>&lt;20</td>
</tr>
<tr>
<td>Cr</td>
<td>wppm</td>
<td>14</td>
<td>12</td>
<td>22</td>
<td>91</td>
</tr>
<tr>
<td>Fe</td>
<td>wppm</td>
<td>108</td>
<td>103</td>
<td>134</td>
<td>392</td>
</tr>
<tr>
<td>Mg</td>
<td>wppm</td>
<td>&lt;39</td>
<td>&lt;40</td>
<td>&lt;40</td>
<td>&lt;40</td>
</tr>
<tr>
<td>Hg</td>
<td>wppm</td>
<td>&lt;0.05</td>
<td>&lt;0.05</td>
<td>&lt;0.05</td>
<td>&lt;0.05</td>
</tr>
<tr>
<td>Ni</td>
<td>wppm</td>
<td>7</td>
<td>5.4</td>
<td>10</td>
<td>38</td>
</tr>
<tr>
<td>Pd</td>
<td>wppm</td>
<td>169</td>
<td>180</td>
<td>174</td>
<td>185</td>
</tr>
<tr>
<td>P</td>
<td>wppm</td>
<td>48</td>
<td>45</td>
<td>&lt;40</td>
<td>&lt;40</td>
</tr>
<tr>
<td>Si</td>
<td>wppm</td>
<td>306</td>
<td>246</td>
<td>325</td>
<td>325</td>
</tr>
<tr>
<td>Ag</td>
<td>wt.%</td>
<td>0.11</td>
<td>0.12</td>
<td>0.11</td>
<td>N/A</td>
</tr>
<tr>
<td>Na</td>
<td>wppm</td>
<td>72</td>
<td>84</td>
<td>68</td>
<td>68</td>
</tr>
<tr>
<td>S</td>
<td>wppm</td>
<td>&lt;17</td>
<td>&lt;17</td>
<td>&lt;17</td>
<td>&lt;17</td>
</tr>
<tr>
<td>V</td>
<td>wppm</td>
<td>3</td>
<td>&lt;3</td>
<td>&lt;3</td>
<td>&lt;3</td>
</tr>
</tbody>
</table>

- Argon tracer studies were done to determine if there were mal-distribution problems within the catalyst bed. This work showed that mal-distribution was not an issue.

**Additional work done to “prove or disprove” the supposition that the poor converter performance was being caused by a “catalyst poison” being in the converter feed.**

A plant test was undertaken to determine if the installation of a “guard layer” on top of the 1st bed of G-58D catalyst would lead to an improved performance of the converter. Sud Chemie recommended using their Ag-on-Alumina guard adsorbent, T-2552A. The composition of this adsorbent is: 6% Ag-on-alumina. Feedback from industrial experience confirmed that this was a “good, general purpose” adsorbent of use in protecting selective hydrogenation catalysts. The “B” reactor was used for this test. See Figure 6 for the converter schematic that illustrates the 1st Bed loading with T-2552A on top of G-58D.

Figures 13 and 14 show the performance of “B” Reactor before and after installation of T-2552A. The operators managed to operate the “B” reactor for about 6 months with the layer of T-2552A. However, the selectivity of the converter was still very poor.

This long cycle of the “B” reactor was the 1st Cycle after commissioning of the analyzer upgrade project. Hence, the ability to detect and react more quickly to trace acetylene leakage was a likely reason for this long cycle.
In addition to this “guard layer trial”, slipstream test facilities were installed for testing a number of other adsorbent traps. This work did not get underway before implementing catalyst and facilities changes that brought about the dramatic improvement in the performance of the converter.

**IN LATTER PART OF 2002, CONVERTER IMPROVEMENT TEAM BEGAN TO FOCUS ON IMPLEMENTING A “THREE-PRONGED SOLUTION”**

This three-pronged approach was as follows:

- Replace G-58D with G-58C
- Install facilities to inject CO into the converter feed to “moderate G-58C”
- Install a 3rd reactor downstream of the current converter system.
  - G-58C would also be used in this reactor.
  - It would be used primarily as a “polishing reactor”.
  - Initially, it would convert small leakages of acetylene from the 2nd beds and it would provide “insurance” such that the operators could tune the 1st and 2nd beds to operate more selectively.
  - This would also provide the additional catalyst volume required as the “acetylene load” increases through higher severity naphtha cracking.

The first cycle with two beds of G-58C was with the “A” reactor. This got underway in May 2003. Figure 6 is a schematic illustrating the loading of G-58C in the “A” reactor. For this run, the layer of C-31-3 was still in the 2nd Bed. Furthermore, the CO injection facilities were not yet commissioned. Nonetheless, the cycle length was more than 3 months, and there was a positive ethylene gain for about 7 weeks. Figure 15 shows the performance trend for this cycle. The ethylene gain during the 1st three weeks averaged about 1000 Lbs/Hr. (Note: For simplification, the formation of C4+ has been ignored. On average, the amount of acetylene downgraded to C4+ material is about 10%.) This cycle was the “first solid indication” that the proposed “ultimate converter system” was going to bring about a big improvement in the performance of the acetylene converter.

In the 4th quarter of 2003, the C-31-3 layer in the 2nd Bed of the “A” reactor was replaced with G-58C. In the 1st quarter of 2004, the G-58D in the “B” reactor was replaced with G-58C and the CO injection facilities were commissioned. Figure 7 is a schematic of the 3-Bed System.

Figures 16 and 17 illustrate the performance of the acetylene converter during 2005 following commission of the CO injection facilities and the 3rd Bed. Figure 16 illustrates
the performance of “A” reactor and Figure 17 illustrates the performance of the “B” reactor.

Some general comments about the performance of the converter system since late 2003 are as follows:

- No flaring events during reactor switching
- Cycle lengths are 3-4 months. There is still a positive ethylene gain at the time of switch.
- The converter is very stable, and the catalyst beds “respond as expected” to changes made to inlet temperatures and H2 injection rates.
- There has not been an “off-spec” acetylene incident attributable to the performance of the converter.
Figure 1

BASF FINA Petrochemicals Port Arthur Site – Adjacent to Total's Refinery
Figure 2
BASF FINA Petrochemicals LP (Naphtha Cracker) and Sabina Petrochemicals LLC
Figure 3. Block Diagram for NROC Cracker

BASF/FINA PETROCHEMICALS
NAFTA Region Olefins Complex (NROC)
Port Arthur, Texas

Cracking Heaters → Primary Fractionator System → Quench Water System

Ethane/Propane Recycle

Heavy Gasoline

Medium Gasoline → Condensate Stages

Charge Gas Compressor System → Charge Gas Co-wash

Gas/Liquid Dryers & Regeneration System

Charge Gas Chilling Train → Methanator and Hydrogen Dryers

Demethanizer & Methane Compressor

Deethanizer

Debutanizer

Demethanizer → 2nd Stage Pyrolysis Gasoline Hydro.

Recycled C5's

C2/C3 Non-Aromatics

2nd Stage Pyrolysis Gasoline Hydro.

C5+ Hydro for Recycle to Furnaces

C2's to Pipelines

Hydrogen to Hydrotreater & Export

Pipeline Hydrogen

Pipeline Natural Gas

Polymer Grade Ethylene Product to pipeline

Polymer Grade Propylene Product to pipeline

Polymer Grade Benzene

Benzene Product

Toluene Product

Heavy PyGas (C5+) to Total Refinery

Pyrolysis Fuel Oil & Pyrolysis Gas Oil Product to Total Refinery

Naphtha Furnace Feeds

Ethane/Propane Recycle

C1, C2

C5, C6

C7, C8

C2/C3 Splitter

C2 Splitter

C3 Splitter

C4 Splitter

C5 Splitter

C6/C7 Non-Aromatics

C8/C9

Pyrolysis Fuels

Polymers Grade

Ethylene

Propylene

C4's to Sabina (C4 Complex)

Naphtsalene

Furnace Feeds

C2/C3 Splitter

C2 Splitter

C3 Splitter

C4 Splitter

C5 Splitter

C6/C7 Non-Aromatics

C8/C9

Pyrolysis Fuels

Polymers Grade

Ethylene

Propylene

C4's to Sabina (C4 Complex)
Figure 4
NROC Tail-End Acetylene Converter at Start-up in December 2001
Figure 5
NROC Tail-End Acetylene Converter
Illustrating Trial Operation with G-58C/C-31-3 in 2nd Bed of Reactor “A”
Figure 6
NROC Tail-End Acetylene Converter
Illustrating Trial Operation with T-2552 Guard Layer
in 1st Bed of Reactor “B”
Figure 6A
NROC Tail-End Acetylene Converter
Illustrating 1st Cycle with G-58C Catalyst in
1st and 2nd Beds Reactor “A”
Figure 7
Current Configuration of NROC Tail-End Acetylene Converter
Figure 10 - Reactor A Cycle 2 Ethylene Gain

- Actual Ethylene Gain
- Max Potential Ethylene Gain
Figure 11 - Reactor B  Cycle 2  Ethylene Gain

Date
03/15/02 03/25/02 04/04/02 04/14/02 04/24/02 05/04/02 05/14/02 05/24/02 06/03/02

Ethylene Gain, Lbs/Hr
-6000 -5000 -4000 -3000 -2000 -1000 0 1000 2000 3000 4000 5000

Max Potential Ethylene Gain
Acetylene Breakthrough
Actual Ethylene Gain
Plant Unstable or Down

Actual Ethylene Gain  Max potential Ethylene Gain
Figure 12 - Reactor A Cycle 3 Ethylene Gain
2nd Bed Loaded with G-58C(Top 75%) and C-31-3(Btm 25%)
Figure 13 - Reactor B Cycle 3 Ethylene Gain

Ethylene gain, Lbs/Hr

Date

Max Potential Ethylene Gain

Actual Ethylene Gain

06/23/02 07/03/02 07/13/02 07/23/02 08/02/02 08/12/02 08/22/02 09/01/02 09/11/02
Figure 14 - Reactor B  Cycle 4  Ethylene Gain
Guard Layer -T-2552A in 1st bed

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<th>Max Potential Ethylene Gain</th>
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- Actual Ethylene Gain
- Max Potential Ethylene Gain
Figure 15 - Reactor A 1st cycle with G-58C in 1st bed and Mix of G-58C/C-31-2 in 2nd Bed
Without CO Moderation

Date

Ethylene Gain, Lbs/Hr

0 1000 2000 3000 4000

04/28/03 05/19/03 06/08/03 06/28/03 07/18/03 08/07/03 08/27/03 09/16/03

Actual Ethylene Gain

Max Potential Ethylene Gain

Actual Ethylene Gain

Max Potential Ethylene Gain
Figure 16 - Reactor A - 1st Cycle with G-58C in Beds 1 & 2
With CO Injection and 3rd Bed in Service
Figure 17 - Reactor B 2nd Cycle with G-58C in Beds 1 & 2
With CO Injection and 3rd Bed In Service

![Graph showing ethylene gain over time with labels for max potential ethylene gain, actual ethylene gain, and plant upsets.](image-url)